

Match level:

1:CLASS2:CLASS3:CLASS4:CLASS5:CLASS6:CLASS7:CLASS8:CLASS9:CLASS

10:CLASS11:CLASS12:CLASS13:CLASS

fragments assigned product role:

containing 5

fragments assigned reactant/reagent role:

containing 1

* * * * * * * * * * * * * * * * STN Columbus * * * .* * * * * * * * * * * * * * *

FILE 'HOME' ENTERED AT 09:54:39 ON 23 FEB 2007

=> file casreact

SINCE FILE TOTAL ENTRY SESSION 0.21 0.21

COST IN U.S. DOLLARS
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FILE CONTENT: 1840 - 18 Feb 2007 VOL 146 ISS 8

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=>
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L1 STRUCTURE UPLOADED

=> d

L1 HAS NO ANSWERS

L1 STR

*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***

Structure attributes must be viewed using STN Express query preparation.

=> s 11

SAMPLE SEARCH INITIATED 09:55:21 FILE 'CASREACT'
SCREENING COMPLETE - 2497 REACTIONS TO VERIFY FROM

184 DOCUMENTS

100.0% DONE 2497 VERIFIED 3 HIT RXNS 3 DOCS SEARCH TIME: 00.00.02

FULL FILE PROJECTIONS: ONLINE **COMPLETE**

BATCH **COMPLETE** S: 46948 TO 52932

PROJECTED VERIFICATIONS: 46948 TO 52932 PROJECTED ANSWERS: 3 TO 163

L2 3 SEA SSS SAM L1 (3 REACTIONS)

=> d 12 1-3

L2 ANSWER 1 OF 3 CASREACT COPYRIGHT 2007 ACS on STN

RX(1) OF 1

 $H_2C = CH - CH = CH_2$ $H_2C = CH - CH - CH_2$ $H_2C = CH_2$ $H_2C =$

 $H_2C = CH - (CH_2)_3 - CH = CH - CH_2 - OH$

25%

REF: Jpn. Kokai Tokkyo Koho, 09059193, 04 Mar 1997, Heisei

L2 ANSWER 2 OF 3 CASREACT COPYRIGHT 2007 ACS on STN

RX(1) OF 1

H₂C=CH-CH=CH₂ Pd(OAc)2, Tri-o-tolylphosphine, CO2, Water, Me2CO

H₂C== CH- (CH₂)₃- CH== CH- CH₂- OH

REF: Jpn. Kokai Tokkyo Koho, 06287156, 11 Oct 1994, Heisei

L2 ANSWER 3 OF 3 CASREACT COPYRIGHT 2007 ACS on STN

RX(2) OF 2

 H_3C-OH $\frac{1,3-Butadiene}{C:148800-96-8, PPh3}$

 $H_2C = CH - (CH_2)_3 - CH = CH - CH_2 - OMe +$

OMe
H₂C== CH-CH-(CH₂)₃-CH== CH₂

H₃C== CH-CH-(CH₂)₃-CH== CH₂

H₃C==

REF: Jpn. Kokai Tokkyo Koho, 04327594, 17 Nov 1992, Heisei

NOTE: 60.degree. under N2

=> s 11 ful

100.0% DONE

FULL SEARCH INITIATED 09:56:07 FILE 'CASREACT'
SCREENING COMPLETE - 54480 REACTIONS TO VERIFY FROM 3626 DOCUMENTS

180 HIT RXNS 68 DOCS

SEARCH TIME: 00.00.07

L3 68 SEA SSS FUL L1 (180 REACTIONS)

54480 VERIFIED

=> d 13 1-68

RX(1) OF 11

REF: Journal of Molecular Catalysis A: Chemical, 238(1-2), 199-206; 2005

NOTE: chemoselective, regioselective, stereoselective, 30:70 alpha:beta, 8:1 E:Z, optimization study, optimized on amount of butadiene, optimized on reaction time, product depends on reaction conditions

CON: 70 hours, 20 deg C

L3 ANSWER 2 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

stereoisomers

REF: Carbohydrate Research, 2006, 341(1), 153-159; 2005

NOTE: 99% overall yield, alternative reaction conditions gave lower yield, autoclave used, optimization study CON: 24 hours, 75 deg C

L3 ANSWER 3 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

$$H_2C$$
 $(CH_2)_3$ AcQ O OAC

REF: Green Chemistry, 7(4), 219-223; 2005 NOTE: stereoselective, 83% conversion, Et3N and DIPEA gave lower conversion, optimization study, optimized on base, reactant assumed in 2nd stage

CON: 45 minutes, 80 deg C

ANSWER 4 OF 68 CASREACT COPYRIGHT 2007 ACS on STN L3

RX(1) OF 1

O | 1. Ph2-pentadienone Pd,

$$HO-C-CH_3$$
 \xrightarrow{PhMe} $\xrightarrow{2.1,3-Butadiene}$

Aco-
$$CH_2$$
- CH = CH - $(CH_2)_3$ - CH = CH_2 + H_2C = CH - CH - $(CH_2)_3$ - CH = CH_2

REF: U.S. Pat. Appl. Publ., 2005038305, 17 Feb 2005

NOTE: optimization study, regioselective

STAGE(1) 0.17 hours, 25 deg C; 25 deg C -> -60 deg C STAGE(2) 215 deg C; 8 hours, 25 deg C -> 60 deg C

ANSWER 5 OF 68 CASREACT COPYRIGHT 2007 ACS on STN L3

RX(6) OF 28

$$H_2C = CH - CH = CH_2$$
 $\xrightarrow{\text{MeOH}, C:781673-24-3,}$ $\xrightarrow{\text{NaOMe}}$

$$H_2C = CH - (CH_2)_3 - CH = CH - CH_2 - OMe + 93%$$

$$_{\text{H}_2}$$
C = CH - CH₂ - CH = CH - CH = CH₂ + Me - CH = CH₂ | + Me - CH = CH₂ | + OMe | C - (CH₂)₃ - CH = CH₂ + OMe | + OMe | CH = CH₂ | +

REF: Chemistry--A European Journal, 10(16), 3891-3900; 2004 NOTE: chemoselective, regioselective, alternative catalysts gave lower

yields, stainless steel Parr autoclave used

STAGE(1) -78 deg C; -78 deg C -> 70 deg C; 16 hours, 70 deg C

ANSWER 6 OF 68 CASREACT COPYRIGHT 2007 ACS on STN L3

1. 1,3-Butadiene, PPh3, Pd acetylacetonate, Et3N, DMF

2. Ac2O, Pyridine

stereoisomers

RX(2) OF 8

stereoisomers

European Journal of Organic Chemistry, (13), 2914-2922; 2004 other products also detected, stereoselective, isomer mix; 86% NOTE: overall yield; selectivity depends on time, additive, cat. and

amt. of amine and phosphine CON: STAGE(1) 45 minutes, 75 deg C; cooled

STAGE(2) 12 hours, room temperature

1,3-Butadiene, HO- CH2- CH2- CH2- OH C:63995-70-0, Pd(OAc)2, Water

 $H_2C = CH - (CH_2)_3 - CH = CH - CH_2 - O - (CH_2)_3 - OH$

Advanced Synthesis & Catalysis, 345(11), 1242-1246; NOTE: autoclave used, other product also detected, yield depends on

reaction conditions

4 hours, 80 deg C, 10 bar -> 5 bar

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RX(1) OF 5

stereoisomers

RX(1) OF 5

stereoisomers

REF: European Journal of Organic Chemistry, (3), 511-520; 2004

NOTE: regioselective, stereoselective, 49:51 alpha:beta, 97 % overall

yield;autoclave used;optimization study;optimized on solvent

used;other solvents (DMF, DCM and MeCN) gave lower yield and

regioselectivity;yield proportions determined by GC;;%

4 hours, 70 deg C CON:

ANSWER 9 OF 68 CASREACT COPYRIGHT 2007 ACS on STN L3

RX(3) OF 7

1,3-Butadiene, HO- CH2- CH2- OH Pd acetylacetonate, C:63995-70-0, Water

 $H_2C = CH - (CH_2)_3 - CH = CH - CH_2 - O - CH_2 - CH_2 - OH +$

о- сн₂- сн₂- он н2С== СН-СН- (СН2) 3-СН== СН2

REF: Green Chemistry, 5(2), 198-204; 2003 NOTE: 75 % overall yield; green chem.-renewable feedstock; high pressure; other products also detected; selective to monotelomers; stainless

steel autoclave used%

STAGE(1) room temperature; room temperature -> 80 deg C; 4 hours, CON: 80 deg C, 10 bar

L3 ANSWER 10 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

RX(1) OF 15

MeOH, Pd(OAc)2, PPh3, H₂C== CH- CH== CH₂ S:350493-08-2

H₂C== CH- CH₂- CH₂- CH= CH- CH= CH₂ + HO

OMe H₂C== CH-CH- (CH₂)₃-CH== CH₂

Organometallics, 22(22), 4418-4425; 2003

NOTE: ionic liq., product distribution depends on solvent, ligand,

other products also detected, 100% conversion

STAGE(1) -10 deg C; 1 hour, 85 deg C

ANSWER 11 OF 68 CASREACT COPYRIGHT 2007 ACS on STN L3

RX(3) OF 6

MeOH, Pd complex, H2C= CH- CH= CH2 (p-MeC6H4)3P

 $H_2C = CH - (CH_2)_3 - CH = CH - CH_2 - OMe +$ $H_2C = CH - CH - (CH_2)_3 - CH = CH_2$

European Journal of Organic Chemistry, (2), 274-283; 2003 REF:

NOTE: regioselective, other products also detected (butadiene dimers), optimization study of catalyst

STAGE(1) room temperature; room temperature -> -20 deg C; -20 deg C; .75 hours, 60 deg C, 4 bar CON:

$$H_2C = CH - CH = CH_2$$
 MeOH, C:441018-46-8,

$$_{2}^{\text{C}} = _{\text{CH- (CH}_{2})_{3}} - _{\text{CH- CH- CH}_{2}} = _{\text{CH- CH- (CH}_{2})_{3}} - _{\text{CH- CH$$

REF: Angewandte Chemie, International Edition, 41(6), 986-989; 2002 NOTE: 98% overall, chemoselective

L3 ANSWER 13 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

RX(5) OF 15

HO-
$$(CH_2)_6$$
-OH $\frac{1,3-Butadiene}{Pd(OAc)2, PPh3, H2}$

$$H_2C = CH - (CH_2)_3 - CH = CH - CH_2 - O - (CH_2)_6 - OH$$

REF: Jpn. Kokai Tokkyo Koho, 2002020343, 23 Jan 2002

L3 ANSWER 14 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

RX(1) OF 10

$$\begin{array}{c|c}
O \\
\parallel \\
MeO-C-(CH2)5-OH
\end{array}$$

$$\frac{1,3-Butadiene, PBu3,}{Pd(OAc)2} \Rightarrow$$

REF: Jpn. Kokai Tokkyo Koho, 2001240598, 04 Sep 2001

L3 ANSWER 15 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

$$H_2C = CH - CH = CH_2$$
 $\xrightarrow{Pd(OAc)2, PPh3, Water,}$ $\xrightarrow{CO2, Me2CO}$

$$H_2C = CH - (CH_2)_3 - CH = CH - CH_2 - OH +$$

H2C== CH- CH2- CH2- CH== CH- CH== CH2 +

REF: Journal of Molecular Catalysis A: Chemical, 166(2), 233-242;

2003

NOTE: products and the ratio related cat

L3 ANSWER 16 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

RX(1) OF 1

$$H_2C = CH - CH = CH_2$$
 MeOH, Pd(OAc)2, PPh3,
Triethylenediamine

$$H_2C = CH - (CH_2)_3 - CH = CH - CH_2 - OMe + H_2C$$

$$\begin{array}{c} \text{OMe} \\ | \\ \text{H}_2\text{C} = \text{CH-CH-(CH}_2)}_3 - \text{CH} = \text{CH}_2 \end{array}$$

REF: Advanced Synthesis & Catalysis, 343(1), 29-33; 2001 NOTE: 80% conversion

L3 ANSWER 17 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

RX(5) OF 156

REF: European Journal of Organic Chemistry, (17), 2991-3000; 2000

$$_{\text{H}_2\text{C}} = \text{CH- (CH}_2)_3 - \text{CH} = \text{CH- CH}_2 - \text{OH} + \\ \text{H}_2\text{C} = \text{CH- CH- (CH}_2)_3 - \text{CH} = \text{CH}_2$$

REF: Jpn. Kokai Tokkyo Koho, 11189556, 13 Jul 1999, Heisei NOTE: 75.degree. for 3 h; 1,7-octadien-3-ol/2,7-octadiene-1-ol ratio of 20.3

L3 ANSWER 19 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

RX(1) OF 1

$$H_2C = CH - CH = CH_2$$

$$\begin{array}{c} Pd (OAc) 2, \\ \underline{C:213543-25-0, CO2}, \\ Water, Me2CO \end{array}$$

$$H_2$$
C=CH-(CH₂)₃-CH=CH-CH₂-OH + H_2 C=CH-CH-(CH₂)₃-CH=CH₂

70%

REF: Jpn. Kokai Tokkyo Koho, 10237082, 08 Sep 1998, Heisei

L3 ANSWER 20 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

RX(1) OF 5

$$H_2C = CH - CH = CH_2$$
 MeOH, Pd pentadienone, \rightarrow PEt3, Hexane

$$H_2C = CH - (CH_2)_3 - CH = CH - CH_2 - OMe$$

REF: Journal of Molecular Catalysis A: Chemical, 129(2-3), 179-189;

NOTE: seeking to suppress trimer and dimer formation; excess alc. required, phosphine has little effect

L3 ANSWER 21 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

RX(2) OF 2

$$_{\rm H_3C^-OH}$$
 1,3-Butadiene, $_{\rm Pd\,(OAc)\,2,}$ $_{\rm C:110107-24-9}$ $_{\rm H_2C^-CH^-\,(CH_2)\,3^-CH^-CH_2^-OMe\ +}$

$$\begin{array}{c} \text{CH} = \text{CH}_2 \\ | \\ \text{MeO- CH}_2 - \text{CH} = \text{CH- CH}_2 - \text{CH} = \text{CH- (CH}_2)_3 - \text{CH} = \text{CH}_2 \\ \end{array}$$

MULTI
PAGE
IMAGE
192522-71-7

REF: Jpn. Kokai Tokkyo Koho, 09176051, 08 Jul 1997, Heisei NOTE: heating to 90.degree. over 20 min and at 90.degree. for 30 min

L3 ANSWER 22 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

RX(1) OF 1

$$H_2C = CH - (CH_2)_3 - CH = CH - CH_2 - OH$$
25%

REF: Jpn. Kokai Tokkyo Koho, 09059193, 04 Mar 1997, Heisei

L3 ANSWER 23 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

RX(1) OF 1

$$H_2C = CH - CH = CH_2$$

$$\frac{Pd(OAc)2,}{C:115034-38-3, Et3N,}$$
Water, CO2, Me2CO

$$_{12}^{\text{C}}$$
 CH- (CH₂)₃-CH= CH-CH₂-OH + $_{12}^{\text{OH}}$ CH-CH-(CH₂)₃-CH= CH₂

REF: Ger. Offen., 19547498, 27 Jun 1996

L3 ANSWER 24 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

$$H_2C$$
 CH-CH=CH₂ CH_2 CH_2 CH_2 CH_2 CH_2 CH_2 CH_2 CH_3 CH_4 CH_2 CH_4 CH_4 CH_5 CH_5 CH_5 CH_6 CH_7 CH_8 CH

$$H_2C = CH - (CH_2)_3 - CH = CH - CH_2 - OH + H_2C = CH - CH - (CH_2)_3 - CH = CH_2$$

REF: Ger. Offen., 19523335, 04 Jan 1996

L3 ANSWER 25 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

RX(1) OF 1

$$H_2C = CH - CH = CH_2$$
 $CH = CH_2 =$

REF: PCT Int. Appl., 9530636, 16 Nov 1995

L3 ANSWER 26 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

RX(1) OF 1

$$H_2C = CH - CH = CH_2$$
 $C: 63995 - 75 - 5,$
 $C: 157978 - 20 - 6, Et3N,$
 $Pd(OAc) 2, Water$

REF: PCT Int. Appl., 9526948, 12 Oct 1995

L3 ANSWER 27 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

RX(1) OF 3

HO (CH₂)
$$_3$$
 CH₂ + $_{\text{H}_2\text{C}}$ CH- CH- (CH₂) $_3$ - CH= CH₂

REF: Journal of the Chemical Society, Chemical Communications, (9), 931-2; 1995

NOTE: 27% OVERALL YIELD, NATURE OF CATALYST AND REACTION TEMP. DETERMINE PRODUCT DISTRIBUTION

L3 ANSWER 28 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

RX(1) OF 1

H2C=CH-(CH2)3-CH=CH-CH2-OH

REF: Jpn. Kokai Tokkyo Koho, 06287156, 11 Oct 1994, Heisei

L3 ANSWER 29 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

RX(1) OF 1

H₂C=CH- (CH₂)₃- CH= CH- CH₂- ОН

REF: Jpn. Kokai Tokkyo Koho, 06287155, 11 Oct 1994, Heisei

L3 ANSWER 30 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

RX(1) OF 3

 $H_2C = CH - (CH_2)_3 - CH = CH - CH_2 - OH$

REF: Ger. Offen., 4410746, 06 Oct 1994

L3 ANSWER 31 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

RX(1) OF 1

$$H_2C = CH - CH = CH_2$$
 Pd acetylacetonate, \Rightarrow PPh3, Water

$$_{12}^{\text{C}}$$
 CH- (CH₂)₃-CH= CH-CH₂-OH + $_{12}^{\text{OH}}$ CH-CH-(CH₂)₃-CH= CH₂

REF: PCT Int. Appl., 9400410, 06 Jan 1994 NOTE: CO2 pressure 20 kg/cm2; 90.degree.

L3 ANSWER 32 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

$$H_2C = CH - CH = CH_2$$
 PPh3, C:29335-52-2, Water, CO2, DMF

REF: Jpn. Kokai Tokkyo Koho, 05155795, 22 Jun 1993, Heisei

NOTE: 90.degree., 3 h; 78% butadiene conversion; 74% selectivity for 2,

4-octadienol.

L3 ANSWER 33 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

RX(1) OF 2

REF: Eur. Pat. Appl., 546422, 16 Jun 1993

NOTE: no solvent

L3 ANSWER 34 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

RX(1) OF 1

$$H_2C = CH - CH = CH_2$$
 $CH - CH = CH_2$
 $CH - CH_2$
 $CH - CH = CH_2$
 $CH - CH = CH_2$
 $CH - CH_2$
 CH

$$H_2C = CH - (CH_2)_3 - CH = CH - CH_2 - OH$$

REF: Brit. UK Pat. Appl., 2260136, 07 Apr 1993

L3 ANSWER 35 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

RX(2) OF 2

$$H_3C-OH$$
 1,3-Butadiene, $C:148800-96-8$, PPh3

$$H_2C = CH - (CH_2)_3 - CH = CH - CH_2 - OMe +$$

STRUCTURE

$$\begin{array}{c} \text{OMe} \\ \mid \\ \text{H}_2\text{C} = \text{CH-CH-} (\text{CH}_2)_3 - \text{CH} = \text{CH}_2 \end{array}$$

DIAGRAM IS NOT AVAILABLE

AVALLABLE 26952-74-9

REF: Jpn. Kokai Tokkyo Koho, 04327594, 17 Nov 1992, Heisei NOTE: 60.degree. under N2

L3 ANSWER 36 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

RX(1) OF 1

$$H_3C-OH$$
 $\frac{1.3-Butadiene}{C:304-88-1, Pd, PPh3}$

$$H_2C = CH - (CH_2)_3 - CH = CH - CH_2 - OMe + H_2C = CH - CH_2 - CH_2 - CH_2 + CH_2$$

REF: Jpn. Kokai Tokkyo Koho, 04327552, 17 Nov 1992, Heisei

L3 ANSWER 37 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

RX(1) OF 1

$$H_2C = CH - CH = CH_2$$

Pd acetylacetonate,

PPh3, C:9003-39-8,

CO2, MeCN

$$_{12}^{\text{C}} = _{12}^{\text{CH}} = _{13}^{\text{CH}} = _{14}^{\text{CH}} = _{14}^{\text{CH}} = _{14}^{\text{CH}} = _{14}^{\text{CH}} = _{14}^{\text{C}} = _{14}^{\text{C}}$$

REF: U.S., 5169981, 08 Dec 1992

L3 ANSWER 38 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

$$_{\text{H}_2}$$
C = CH- CH₂- CH₂- CH= CH- CH= CH₂ + 5%

REF: Fenzi Cuihua, 6(2), 148-55; 1992

L3 ANSWER 39 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

HO-CH₂-CH-CH₂-OH Pd acetylacetonate, Me2CHOH

$$\begin{array}{c} \text{OH} \\ | \\ \text{Me- (CH}_2)_7 - \text{O-CH}_2 - \text{CH-CH}_2 - \text{O- (CH}_2)_7 - \text{Me} \end{array}$$

$$\begin{array}{c} \text{O- (CH}_2)_{\,7}\text{-Me} \\ | \\ \text{Me- (CH}_2)_{\,7}\text{-O-CH}_2\text{-CH-CH}_2\text{-O- (CH}_2)_{\,7}\text{-Me} \end{array}$$

REF: Ger. Offen., 4021015, 09 Jan 1992

L3 ANSWER 40 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

$$_{12}^{\text{C}} = _{\text{CH}}^{\text{CH}} = _{\text{CH}}^{\text{CH}} = _{\text{CH}}^{\text{CH}} = _{\text{CH}}^{\text{CH}} = _{\text{CH}}^{\text{OH}} = _{\text{CH}}^{\text{OH}$$

REF: Ger. Offen., 3925217, 31 Jan 1991

NOTE: 93% overall

L3 ANSWER 41 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

RX(10) OF 12

$$H_2C = CH - (CH_2)_3 - CH = CH - CH_2 - OMe +$$

REF: Journal of Molecular Catalysis, 55(1-3), 340-52; 1989

L3 ANSWER 42 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

RX(14) OF 14

 $H_2C = CH - CH = CH_2$ AcOH, Pd(OAc)2, PhMe

Aco-
$$CH_2$$
- CH — CH — CH_2) $_3$ - CH — CH_2 $_4$ $_4$ $_4$ $_2$ C — CH - CH - CH - CH 2) $_3$ - CH — CH 2 $_2$ $_3$ - CH — CH 2 $_2$ $_3$ - CH — CH 2 $_2$ $_3$ - CH — CH 2 $_3$ - CH — CH 2 $_4$ - CH 3 $_4$ - CH 5 $_5$ - CH 5 $_$

REF: Journal of Organic Chemistry, 54(11), 2726-30; 1989

L3 ANSWER 43 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

$$H_2C = CH - CH = CH_2$$
 $C:81141-80-2,$
 $C:63936-88-9, PhMe,$
 $MeOH$

MeO
$$(CH_2)_3$$
 CH_2 +

MeO $(CH_2)_3$ CH_2 +

$$H_2$$
C (CH₂) 3 CH₂

REF: Journal of Organic Chemistry, 54(10), 2459-62; 1989

L3 ANSWER 44 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

RX(1) OF 3

Me Me Me
$$(CH_2)_3$$
 CH_2 73%

REF: Zhurnal Organicheskoi Khimii, 24(1), 119-21; 1988

L3 ANSWER 45 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

RX(6) OF 6-3 STEPS

1. Isopentadiene,
PdC12, PPh3, MeOH

2. ClCO2Me

3. 18-Crown-6, MeCN

REF: Zhurnal Organicheskoi Khimii, 23(11), 2297-9; 1987

L3 ANSWER 46 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

RX(1) OF 4

$$H_2C = CH - (CH_2)_3 - CH = CH - CH_2 - OH +$$

$$\begin{array}{c} \text{OH} \\ \text{H}_2\text{C} = \text{CH} - (\text{CH}_2)_3 - \text{CH} = \text{CH} - \text{CH}_2 - \text{CH} = \text{CH} - (\text{CH}_2)_3 - \text{CH} = \text{CH}_2 \\ \end{array} \\ + \\ \begin{array}{c} \text{CH} + (\text{CH}_2)_3 - \text{CH} = \text{CH}_2 \\ \text{CH} + (\text{CH}_2)_3 - \text{CH} = \text{CH}_2 \\ \end{array}$$

$$\begin{array}{c} \text{CH}_2\text{--}\text{CH} = \text{CH}_-\text{(CH}_2)_3\text{--}\text{CH} = \text{CH}_2 \\ \text{H}_2\text{C} = \text{CH}_-\text{(CH}_2)_3\text{--}\text{CH} = \text{CH}_-\text{CH}_2\text{--}\text{CH} = \text{CH}_-\text{(CH}_2)_3\text{--}\text{CH} = \text{CH}_2 \\ \end{array}$$

REF: Izvestiya Akademii Nauk SSSR, Seriya Khimicheskaya, (10), 2254-6;

1986

NOTE: overall yield 51%

L3 ANSWER 47 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

RX(15) OF 20

$$H_2C = CH - (CH_2)_3 - CH = CH - CH_2 - O - C - (CH_2)_3 - NH_2$$
848

REF: Zhurnal Organicheskoi Khimii, 22(8), 1610-19; 1986

L3 ANSWER 48 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

Ets-CH₂-CH₂-OH

1,3-Butadiene,
Pd acetylacetonate,
PPh3, AlEt3, PhMe

REF: Zhurnal Organicheskoi Khimii, 22(8), 1591-7; 1986

L3 ANSWER 49 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

RX(1) OF 14

Me-NH-Ph Isopentadiene, MeOH, > Pd(OAc)2, PPh3

RX(1) OF 14

REF: Izvestiya Akademii Nauk SSSR, Seriya Khimicheskaya, (6), 1344-7;

1986

NOTE: or Pd(acac)2

L3 ANSWER 50 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

REF: Journal of Organometallic Chemistry, 309(1-2), 215-23; 1986

L3 ANSWER 51 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

RX(1) OF 2

REF: U.S., 4642392, 10 Feb 1987

L3 ANSWER 52 OF 68 CASREACT COPYRIGHT 2007 ACS on STN

RX(1) OF 5

$$^{\text{H}_2\text{C}}_{\parallel}$$
 OH Me $^{\text{CH}_2}_{\parallel}$ He-C-CH₂-CH₂-CH-CH=CH₂ + Me-C-(CH₂)₃-C-CH=CH₂ + OH

Me
$$(CH_2)_3$$
 OH

stereoisomers

REF: Zeitschrift fuer Chemie, 25(6), 226-7; 1985

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$$H_3C-OH$$
 $\frac{1,3-Butadiene}{C:80191-14-6}$, Et20 MeO (CH₂) $\frac{1}{3}$ CH₂ stereoisomers

REF: Organometallics, 5(3), 473-81; 1986

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$$H_2C = CH - CH = CH_2$$
 AcoH, C:2622-08-4, C:99632-71-0, MeCN

Aco
$$(CH_2)_3$$
 CH_2 + $H_2C = CH - CH - (CH_2)_3 - CH = CH_2$ stereoisomers

REF: Organometallics, 5(3), 514-18; 1986

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$$\begin{array}{c|c} & \text{CH}_2 \\ \parallel & \\ \text{H}_3\text{C--C-CH----} \text{CH}_2 \end{array} \qquad \begin{array}{c} \underline{\text{Pd acetylacetonate,}} \\ \underline{\text{PPh3, MeOH}} \end{array} >$$

REF: Journal fuer Praktische Chemie (Leipzig), 327(4), 643-8; 1985

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$$o=c=0$$
 $\frac{1,3-Butadiene,}{Pd(OAc)2, PPh3, DMF}$

$$H_2C = CH - CH_2 - CH_2 - CH = CH - CH = CH_2 +$$

 $AcO-CH_2-CH=CH-(CH_2)_3-CH=CH_2+$

REF: Nippon Kagaku Kaishi, (3), 533-6; 1985

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H₂C=CH-CH=CH₂ MeOH, Pd, Benzene

$$H_2C = CH - (CH_2)_3 - CH = CH - CH_2 - OMe +$$

$$_{12}^{\text{CM}} = _{12}^{\text{CH}} = _{12}^{\text{CH}} = _{13}^{\text{CH}} = _{13}^{\text{CH}$$

REF: Journal of Molecular Catalysis, 29(1), 99-104; 1985

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$$\begin{array}{c|c} & \text{CH}_2 \\ \parallel & \\ \text{H}_3\text{C--C-CH} \longrightarrow \text{CH}_2 \end{array} \qquad \begin{array}{c} \underline{\text{MeOH, PBu3, NaOMe,}} \\ \text{Me2CHOH} \end{array} >$$

REF: Journal fuer Praktische Chemie (Leipzig), 326(5), 729-36; 1984

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 $H_2C = CH - CH = CH_2$ NaOPh PhO-CH₂-CH = CH-(CH₂)₃-CH = CH₂

REF: U.S.S.R., 892851, 15 Jul 1982

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 $H_2C = CH - CH = CH_2$ Phot $CH_2 - CH = CH - (CH_2)_3 - CH = CH_2$

REF: Angewandte Chemie, 94(10), 796-7; 1982

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1,3-Butadiene, PPh3,

REF: Zhurnal Organicheskoi Khimii, 18(1), 46-52; 1982

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 $HO-CH_2-CH=CH-CH_2-OH$ 1,3-Butadiene

REF: Izvestiya Akademii Nauk SSSR, Seriya Khimicheskaya, (8), 1837-42; 1981

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$$H_2C = CH - (CH_2)_3 - CH = CH - CH_2 - O O$$
 $| | | Me - CH - C - Me$

REF: Tetrahedron Letters, 21(39), 3787-90; 1980

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 $H_2C = CH - CH = CH_2$ $\xrightarrow{\text{MeCH2CH2OH, AlEt3,}}$

 $H_2C = CH - (CH_2)_3 - CH = CH - CH_2 - O - CH_2 - CH_2 - OMe$

REF: Zhurnal Organicheskoi Khimii, 16(6), 1157-61; 1980

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octadienyl esters

$$H_2C$$
 $(CH_2)_3$ O OBu-n

REF: Journal of Organometallic Chemistry, 137(3), 309-14; 1977

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$$H_2C = CH - CH = CH_2$$
 CO2, Me2CO HO (CH2)3 CH_2 +

REF: Journal of the Chemical Society [Section] D: Chemical

Communications, (7), 330; 1971 NOTE: Classification: C-Alkylation; Addition; Hydroxylation; # Conditions: CO2 Pd complex acetone; 90 deg 2h; # Comments: 698 yield of 1-OH product, 13% yield of alkene; 10% yield of

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H₂C=CH-CH=CH₂ AcOH, Pd acetylacetonate, Me2NCH2CH2OH, PPh3

Aco
$$(CH_2)_3$$
 CH_2 + $H_2C = CH - CH - (CH_2)_3 - CH = CH_2$

REF: Tetrahedron Letters, (43), 3817-20; 1970

NOTE: Classification: C-Alkylation; Regioselective; Addition;
Acetoxylation; # Conditions: Me2NCH2CH2OH; PPh3 Pd(acac)2; #
Comments: 71% yield of 1-OAc product, 21% yield of 3-OAc product;
8% yield of alkene product

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L3

OH
$$\frac{1,3-\text{Butadiene, PdC12}}{\text{OPh}}$$
 H₂C (CH₂) $\frac{1}{3}$ OPh +

REF: Journal of the American Chemical Society, 89(25), 6793-4; 1967 NOTE: Classification: Dimerisation; Addition; O-Alkylation; # Conditions: PdCl2 PhONa; 100 deg; # Comments: 91% of trans product